

Question	Scheme	Marks	AOs
<b>11(a)</b>	$339 = ab^{20}, 414 = ab^{60} \Rightarrow \frac{414}{339} = b^{40} \Rightarrow b = \sqrt[40]{\frac{414}{339}}$	M1	3.1a
	$b = 1.005$	A1	1.1b
	$339 = ab^{20} \Rightarrow a = \frac{339}{b^{20}}$ or $414 = ab^{60} \Rightarrow a = \frac{414}{b^{60}}$	M1	1.1b
	$a = 307$	A1	1.1b
		<b>(4)</b>	
<b>(b)(i)</b>	$a$ is the concentration in 1960	B1	3.4
<b>(b)(ii)</b>	$b$ is the factor by which the concentration increases each year	B1	3.4
		<b>(2)</b>	
<b>(c)</b>	$450 = 307 \times 1.005^t \Rightarrow 1.005^t = \frac{450}{307}$	M1	3.4
	$1.005^t = \frac{450}{307} \Rightarrow t = \log_{1.005} \frac{450}{307}$ or $\frac{\ln \frac{450}{307}}{\ln 1.005}$	M1	1.1b
	$t = 76.67\dots$	A1	1.1b
	2036 or 2037	A1	3.2a
		<b>(4)</b>	

**(10 marks)****Notes****(a)**M1: Forms 2 equations in  $a$  and  $b$  and solves to obtain a value for  $b$ .A1:  $b = 1.005$ M1: Uses either equation and their value for  $b$  to find a value for  $a$ A1:  $a = 307$ **(b)(i)**B1: Correct interpretation for the constant  $a$ **(b)(ii)**B1: Correct interpretation for the constant  $b$ **(c)**M1: Uses their values of  $a$  and  $b$  and the 450 in the equation for the model and reaches an equation of the form " $1.005^t = k$ "M1: For using correct log work to obtain a value for  $t$ 

A1: For awrt 76.7

A1: Interprets the value of  $t$  correctly and states the year 2036 or 2037, following correct work